Synthesis of Polyfunctionalized Pyrroles by a PPh₃-Promoted Condensation Reaction between Ammonium Acetate, Dialkyl Acetylenedicarboxylate and Arylglyoxals

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ABSTRACT: A simple and efficient synthesis of some polyfunctionalized pyrrole derivatives by a triphenylphosphine-promoted condensation reaction between dialkyl acetylenedicarboxylates, arylglyoxals, and ammonium acetate is described. This present method carries several advantages, such as good yields, a simple procedure, non-hazardous reaction conditions and starting from easily accessible substrates.

KEYWORDS: *Dialkyl acetylene dicarboxylates; Triphenylphosphine; Ammonium acetate; Intramolecular Wittig reaction; Arylglyoxals.*

INTRODUCTION

N-Heterocycles receive considerable attention in the literature as a consequence of their exciting biological properties and their role as pharmacophores [1]. Of these heterocycles, the pyrrole ring is one of the most fundamental. It is a widely distributed structural unit in a variety of natural and biologically important molecules such as porphyrins, bile pigments, coenzymes, and alkaloids [2]. Therefore, it is not surprising that many methods for the syntheses of substituted and functionalized pyrroles have been reported in the literature [3]. A diverse range of pharmacological properties, including antibacterial, antitumor, anti-inflammatory, antioxidant, antianginal and antifungal activities of this important class of heterocycles has been reported in the literature [4].

Recently, syntheses of polysubstituted pyrroles have been reported from conjugate addition reactions [5], transition metal intermediates [6], reductive coupling [7], aza Wittig reactions [8], isocyanide-based reactions [9], utilizing the sila-Stetter/Paal-Knorr sequence strategy [10] and other pathways [11]. However, some of these methods have some drawbacks, such as harsh reaction conditions, lengthy reaction times, expensive catalysts and low yields. Therefore, it is clearly evident that developing new and flexible methods of synthesis is required.

Addition reaction between phosphines or triphenylphosphit and activated carbon-carbon triple bonds is well known to produce a reactive zwitterionic intermediate, which may be trapped by various electrophiles [11-17]. The reaction of triphenylphosphine with dialkyl acetylenedicarboxylates (DAAD) has been studied in the presence of a variety of organic acidic compounds, in order to trap the zwitterionic intermediate.

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Scheme 1: Synthesis of some α -aminophosphorous ylide.

Trapping of PPh₃-DAAD zwitterion by an organic acidic compound containing a carbonyl group has been used as a one-pot and efficient route for the synthesis of a variety of heterocyclic and carbocyclic compounds[16-19]. Treatment of triphenylphosphine with DAAD in the presences of ammonium acetate has been reported to produce α -aminophosphorous ylide **4** [20- 22] (Scheme 1).

Keeping in mind the biological importance of pyrrole ring and in continuation of our current studies on the development of new routes in heterocyclic synthesis [18, 19, 23], herein we report a very simple and highly efficient one-pot method for the synthesis of polyfunctionalized pyrrole derivatives through the reaction of dialkyl acetylene dicarboxylates, ammonium acetate and arylglyoxals in the presence of triphenylphosphine under catalyst-free conditions. (Scheme 2).

EXPERIMENTAL SECTION

Melting points were determined with an electrothermal 9100 apparatus. Mass spectra were recorded on a FINNIGAN-MAT 8430 mass spectrometer operating at an ionization potential of 70 eV. IR spectra were recorded on a Shimadzu IR-470 spectrometer. ¹H and ¹³C NMR spectra were recorded on Bruker DRX-500 Avance spectrometer at the solution in CDCl₃ using TMS as an internal standard. The chemicals used in this work purchased from Fluka (Buchs, Switzerland) and were used without further purification.

General procedure

To a magnetically stirred solution of PPh₃ (1 mmol) and, ammonium acetate (1 mmol) in CH₃CN (10 mL) was added dropwise a mixture of DAAD (1 mmol) in CH₃CN (3 mL) at room temperature over 2 min. The reaction mixture was then stirred for 20 min. After completion of the reaction (TLC), a mixture of

arylglyoxals (1 mmol) in CH₃CN (3 mL)) was added and the reaction mixture was stirred for more 24 h. The solvent was evaporated and the residue was purified by column chromatography on SiO₂ using EtOAC-hexane (1:4) mixture as eluent.

Dimethyl 4-phenyl-1H-pyrrole-2, 3-dicarboxylate (6a)

Yellow crystals, M.p. 135°C. IR (KBr) (v_{max} , cm⁻¹): 3275(NH), 1669, 1729(C=O). ¹H NMR (500.1 MHz, CDCl₃) δ = 3.92 (3 H, s, OCH₃), 3.96 (3 H, s, OCH₃), 6.70-7.60(6 H, m, C₆H₅ and CH), 9.67(1H,br,NH) ppm. ¹³C NMR (125.7 MHz, CDCl₃): δ = 51.95 (OCH₃), 52.34 (OCH₃), 110.71, 121.43, 122.63, 124.90, 128.33, 129.09, 130.30, 135.05 (C arom), 160.93 (CO₂Me), 164.35 (CO₂Me). MS (m/z, %): 259 (M⁺, 9). Anal. Calcd. for C₁₄H₁₃NO₄ : C, 64.86; H, 5.05; N, 5.40 %. Found: C, 64.64; H, 5.21; N, 5.52 %.

Diethyl 4-phenyl-1H-pyrrole-2,3-dicarboxylate (6b)

Yellow crystals, M.p. 141°C. IR (KBr) (v_{max} , cm⁻¹): 3280(NH), 1681, 1729 (C=O). ¹H NMR (500.1 MHz, CDCl₃) δ = 1.19 (3H, t, ³*J*_{HH} = 7.2 Hz, OCH₂CH₃),1.35 (3H, t, ³*J*_{HH} = 7.2 Hz OCH₂CH₃), 4.23 (2H, q, ³*J*_{HH} = 7.2 Hz, OCH₂CH₃), 4.34 (2H, q, ³*J*_{HH} = 7.2 Hz, OCH₂CH₃), 7.01 (1 H, s, CH), 7.33-7.39 (5H, m, arom), 9.46(1H, br, NH) ppm. ¹³C NMR (125.7 MHz, CDCl₃): δ = 14.35, 14.61 (2OCH₂CH₃), 61.42, 61.64 (2OCH₂CH₃), 118.61, 120.43, 123.46, 123.67, 128.12, 129.22, 130.30, 133.82 (C arom), 160.71 (CO₂Et), 164.25 (CO₂Et). MS (m/z, %): 287(M⁺, 11). Anal. Calcd. for C₁₆H₁₇NO₄ : C, 66.89; H, 5.96; N, 4.88%. Found: C, 66.74; H, 5.69; N, 4.96%.

Dimethyl 4-p-tolyl-1H-pyrrole-2, 3-dicarboxylate (6c)

Yellow oil, IR (KBr) (v_{max} , cm⁻¹): 3295(NH), 1713(C=O). ¹H NMR (500.1 MHz, CDCl₃) δ = 2.49 (3H, s, CH₃), 3.86 (3 H, s, OCH₃), 3.89 (3 H, s, OCH₃),



Scheme 2: One-pot synthesis of some functionalized pyrrole derivatives.

6.87-7.57(5 H, m, arom and CH), 9.76 (1H, br, NH) ppm. ¹³C NMR (125.7 MHz, CDCl₃): δ = 21.15(CH₃), 52.15 (OCH₃), 52.41 (OCH₃), 120.04, 121.13, 126.33, 127.29, 128.43, 129.24, 130.20, 130.46 (C arom), 160.61 (CO₂Me), 165.35 (CO₂Me). MS (m/z, %): 273 (M⁺, 20). Anal. Calcd. for C₁₅H₁₅NO₄: C, 65.92; H, 5.53; N, 5.13%. Found: C, 66.08; H, 5.41; N, 5.36%.

Diethyl 4-(4-chlorophenyl)-1H-pyrrole-2,3-dicarboxylate (6d)

Yellow crystals, M.p. 137-139°C. IR (KBr) (v_{max} , cm⁻¹): 3215 (NH), 1690, 1712 (C=O). ¹H NMR (500.1 MHz, CDCl₃) δ =1.20 (3H, t, ³J_{HH} = 7.1 Hz, OCH₂CH₃),1.32 (3H, t, ³J_{HH} = 7.1 Hz OCH₂CH₃), 4.18 (2H, q, ³J_{HH} = 7.1 H Hz, OCH₂CH₃), 4.32 (2H, q, ³J_{HH} = 7.1 Hz, OCH₂CH₃), 7.01 (1 H, s, CH), 7.33-7.39 (4H, m, 4-ClC₆H₄), 9.71(1H, br, NH) ppm. ¹³C NMR (125.7 MHz, CDCl₃): δ = 14.34, 14.49 (20CH₂CH₃), 61.68, 62.10 (20CH₂CH₃), 122.08, 124.86, 125.98, 129.42, 130.18, 132.86, 133.87, 140.72 (C arom), 159.89 (CO₂ Et), 166.13 (CO₂Et). MS (m/z, %): 321 (M⁺, 15). Anal. Calcd. for C₁₆H₁₆ClNO₄ : C, 59.73; H, 5.01; N, 4.35%. Found: : C, 59.52; H, 5.27; N, 4.21%.

Dimethyl4-(4-chlorophenyl)-1H-pyrrole-2,3-dicarboxylate (6e)

Yellow oil, IR (KBr) (ν_{max} , cm⁻¹): 3225(NH), 1668, 1735(C=O). ¹H NMR (500.1 MHz, CDCl₃) δ = 3.86 (3 H, s, OCH₃), 3.88 (3 H, s, OCH₃), 7.06-7.43(5 H, m, arom and CH), 8.92 (1H, br, NH) ppm. ¹³C NMR (125.7 MHz, CDCl₃): δ = 52.69 (OCH₃), 53.04 (OCH₃), 123.50, 124.91, 125.23, 125.87, 127.39, 129.40, 131.38, 133.99 (C arom), 160.38 (CO₂Me), 166.40 (CO₂Me). MS (m/z, %): 293 (M⁺, 11). Anal. Calcd. for C₁₄H₁₂ClNO₄ : C, 57.25; H, 4.12; N, 4.77%. Found: : C, 57.11; H, 4.32; N, 4.55%.

Dimethyl 4-(4-bromophenyl)-1H-pyrrole-2,3dicarboxylate (6f)

Yellow crystals, M.p. 136-138°C. IR (KBr) (v_{max} , cm⁻¹): 3275(NH), 1685, 1745(C=O). ¹H NMR (500.1 MHz, CDCl₃) δ = 3.79 (3 H, s, OCH₃), 3.85 (3 H, s, OCH₃), 7.27-7.80(5 H, m, arom and CH), 9.17(1H, br, NH) ppm. ¹³C NMR (125.7 MHz, CDCl₃): δ = 52.33 (OCH₃), 52.83 (OCH₃), 110.39, 122.33, 129.23, 130.42, 132.22, 132.42, 133.40, 143.40 (C arom), 161.13 (CO₂Me), 170.38



Scheme 3: The suggested mechanism for the formation of functionalized pyrrole derivatives.

(CO₂Me). MS (m/z, %): 336 (M⁺, 29). Anal. Calcd. for $C_{14}H_{12}BrNO_4$: C, 49.73; H, 3.58; N, 4.14%. Found: C, 49.95; H, 3.31; N, 4.32%.

RESULTS AND DISCUSSION

The structures of compounds **6a-f** were deduced from their elemental analyses and their IR, ¹H NMR, ¹³C NMR spectra. For example, the mass spectrum of **6a** displayed the molecular-ion peak at m/z = 259. The 500.1 MH_Z ¹H NMR spectrum of **6a** exhibited two sharp signals at δ 3.92 and 3.96 ppm for two methoxy groups' protons. The aromatic protons were observed at 6.70-7.60 ppm. A broad singlet was observed at δ 9.67 ppm for NH proton. The ¹³C NMR spectrum of compound **6a** showed 12 distinct resonances in agreement with the proposed structure. The structural assignments made on the basis of the NMR spectrum. NH group showed an absorption band at 3275 cm⁻¹ and carbonyl groups exhibited strong absorption bands at 1729, 1669 cm⁻¹.

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A mechanistic rationalization for the reaction is given in Scheme 3. On the basis of the well-established chemistry of trivalent phosphorus nucleophiles [23-28], it is reasonable to assume that the initial event is the formation of the zwitterion 7 from the triphenylphosphine and the acetylenic ester. Next, the zwitterion is protonated by ammonium acetate. The resulting positively charged phosphonium ion 8 is attacked by the conjugate base of NH₃, leading to α -aminophosphorous ylide 4 [20, 21, 22]. which then reacted with arylglyoxals 5 to produce intermediate 9 that underwent intramolecular Wittig reaction and then loses a molecule of water and aromatizes to product 6 under reaction condition (Scheme 3).

CONCLUSIONS

In conclusion here we report the reaction between dialkyl acetylene dicarboxylates, ammonium acetate and arylglyoxals promoted by triphenylphosphine, to produce functionalized pyrrole derivatives in high yields. The present method carries the advantage that not only is the reaction performed under simple conditions but also that the substances can be mixed without any activation or modification.

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