# Experimental and Numerical Study of CO<sub>2</sub>/CH<sub>4</sub> Separation Using SAPO-34/PES Hollow Fiber Membrane

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**ABSTRACT:** In this work, the defects in poletherysulfone (PES)/silicoaluminophosphate (SAPO)-34 zeolite mixed matrix membrane was prepared by dry-wet spinning technique for the separation of  $CO_2/CH_4$  mixtures. In this regard, the synthesized PES/SAPO-34 Mixed Matrix Membranes (MMMs) were characterized via FESEM analyses. The Response Surface Methodology (RSM) was applied to find the relationships between several explanatory variables such as air gap distance, jet stretch ratio, and zeolite content, and  $CO_2$  permeance as responses. The results were validated with the experimental data, which the model results were in good agreement with the available experimental data. The effects of feed temperature and feed pressure on permeation and  $CO_2/CH_4$  selectivity of membranes were investigated. The MMMs showed better performance than the neat PES membrane. A two-dimensional countercurrent mathematical model for membrane separation has been incorporated with Aspen HYSYS to optimize and design the membrane system for  $CO_2$  capture from natural gas. Permeation results manifested that the PES/SAPO-34 fabricated at optimum conditions has incredible worth from the perspective of industrial separations of  $CO_2$  from the flue and natural gas.

**KEYWORDS:** Hollow fiber; Membrane; SAPO 34; Polyethersulfone; CO<sub>2</sub> Separation.

#### INTRODUCTION

Carbon dioxide (CO<sub>2</sub>) is the largest concern of global warming in the past five years [1-3]. CO<sub>2</sub> is in natural gas streams, flue gas from fossil fuel combustion, and a product of coal gasification. The presence of CO<sub>2</sub> and other acid gases reduce the calorific value and make the gas streams become acidic and corrosive, which decrease gas compression and transport. Pipeline specifications usually need CO<sub>2</sub> concentrations below 2% [2]. One major step to treat the various gas streams is to remove acid gas such as CO<sub>2</sub>, H<sub>2</sub>S, and SO<sub>2</sub> before it is compressed and delivered[6]. Highly concentrated CO<sub>2</sub> can be produced from such separation processes rather than direct release

into atmosphere. By pumping and storing pure or highly concentrated  $CO_2$  deep underground, sequestration may provide one feasible approach to treating greenhouse gas emissions [4]. Therefore, economic and effective techniques for  $CO_2$  removal from  $CH_4$  and capture at a wide range of  $CO_2$  concentration levels and flow rates are highly desirable and have attracted great interest [5].

The low capital and operating cost of membrane for removal of impurities from natural gas streams is growing rapidly. Gas permeation characteristics of PES-SAPO-34-HMA were investigated by *Elif et al.*[6]. The results proved that the permeability of all the gases through

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PES-SAPO-34-HMA membranes was significantly more than those through PES/HMA membranes. New MMMs based on polymerizable room-temperature ionic liquids and SAPO-34 as filler were studied [7]. It was shown that increasing of ionic liquids in the MMMs, increases the CO<sub>2</sub> permeability. The effect of the addition of SAPO-34 as filler on the gas permeation properties of other polymers such polysulfone [8] has been researched. Gas separation studies in the field of adding SAPO-34 particles into the polyurethane matrix have been not reported before. The effect of silica particles on the permeation properties of polyurethane membranes was investigated [9]. The obtained results indicated the reduction in permeability of CO2, CH4, O2 and N<sub>2</sub> gases. But enhancement of CO<sub>2</sub>/CH<sub>4</sub>, CO<sub>2</sub>/N<sub>2</sub>and O<sub>2</sub>/N<sub>2</sub> selectivity by increasing silica content was observed. Lots of review papers provide a better understanding of the incorporation of inorganic fillers in MMMs' and the outlook of fillers for  $CO_2/CH_4$  and  $CO_2/N_2$  separation [10].

In this research, the gas separation properties of PES-SAPO 34 hollow fiber MMMs were explored. The hollow fiber membranes were spun from polymer solution containing PES/modified SAPO 34 at various zeolite loading, air gap distance and dope extrusion rate. In order to verify the effect of these parameters CO<sub>2</sub> and CH<sub>4</sub> permeance, response surface on methodology (RSM) based on central composite design (CCD) was used. Recently, response surface method (RSM) is a mathematical and statistical method employed for modeling and optimization of numerous processes. Central Composite Design (CCD) in the RSM is a main design tool used for optimization of methods. The CCD provides complete results and detailed information even for a small number of experiments and positive effects of operating parameters on all responses. The effects of operational parameters such as pressure, temperature and gas composition on permeability (CO<sub>2</sub> and CH<sub>4</sub> gases) and selectivity of those membranes were further studied. The paper also demonstrates the numerical study of CO<sub>2</sub> removal from natural gas through adaptation of proposed hollow fiber membrane module to compare the performance of different flow configurations based on their separation efficiency and process economics. As membrane unit is not a pre-defined unit operation in Aspen Plus, membrane model is included in the process simulation as user defined unit operation along with other available unit operations using A FORTRAN sub-routine.

# MATERIALS AND METHODS

MMMs were prepared using PES as polymer matrix due to its excellent thermal and mechanical characteristics. SAPO-34 is a proven facilitator for better CO2/CH<sub>4</sub> separation. CO<sub>2</sub> gas (99.99% purity) was purchased from Farafan Gas Co. (Isfahan, Iran) and CH<sub>4</sub> gas was acquired from Air Products Co. (Tehran, Iran). Other chemicals were taken from Sigma Alderich (USA).

# Membrane preparation

SAPO-34 zeolite was synthesized via hydrothermal method [11]. Asymmetric porous SAPO-34/PES hollow fiber membranes were prepared by the dry–wet spinning technique according to the phase inversion process. Bore fluid was prepared by dissolving SAPO-34 nanoparticles in solvent followed by gentle stirring for 24 h until a homogenous solution was attained. Hollow fibers were made by the solution spinning technique. The bore fluid was extruded through the spinneret using ISCO syringe pump. In the next step, dope solution was pumped to the spinneret under the N<sub>2</sub> atmosphere. Hollow fibers were kept in water for at least 24 hours and then in aqueous ethanol solutions with ethanol for 15min. The jet–stretch (JS) ratios of the spin line were manipulated by adjusting the take-up rate. The JS ratio is defined as the following equations.

$$JS = \frac{V_2}{V_1}$$
(1)

 $V_1$  is the linear extrusion rate of the extruded fiber (cm/s), while  $V_2$  is the rate at which the fiber is collected at the end of the spin line i.e. take-up rate (cm/s). Fig. 1 shows the schematic of experimental setup. Spinning parameters are listed in Table 1.

# Membrane characterization

The thermal stability of polyurethane MMMs was studied by thermogravimetric analysis (TGA) within the temperature range of 25–850 °C at 10 °C·min<sup>-1</sup>. The morphology of MMMs was investigated using a FE-SEM, Hitachi S-900. Differential scanning calorimetry (DSC) analysis was performed using a PerkinElmer DSC 8000 calorimeter.

# Gas Separation Performance

The constant-volume apparatus (Aramis CS1325, Iran) was used to test the membranes permeabilities. The membrane module was sealed with gasket in a stainless steel cell



Fig. 1: Hollow fiber membrane spinning system: (1) nitrogen cylinder; (2) dope reservoir; (3) gear pump; (4) on-line filter, 7 mm;
(5) syringe pump; (6) spinneret; (7) forced convective tube; (8) roller; (9) wind-up drum; (10) refrigeration/heating unit;
(11) coagulation bath; (12) washing/treatment bath; (13) wind-up bath; and (14) schematic spinneret.

which was fixed inside an oven. The feed pressure of gases in feed stream was controlled by pressure sensors (2.600 G BD pressure sensors, accuracy of 0.25 FSO and DMP 343 BD pressure sensor, accuracy of 0.175% FSO, 1 mbar) and flow rates of gases were controlled using mass flow controllers (MKS Instruments). Temperature sensor (PT-100 sensors, accuracy of 0.1 K) was used. CO<sub>2</sub> and CH<sub>4</sub> permeation of the membranes were measured at feed flow rates of the gases 200 ml/min. The pressure-normalized flux or permeance, ( $P_i/l$ ) was calculated by the equation:

$$\frac{P_i}{l} = \frac{N}{\Delta P} = \frac{Q}{A \times \Delta P} \left( \frac{273.15 \times 10^6}{T} \right)$$
(2)

$$\alpha_{ij} = \frac{\frac{P_i}{l_i}}{\frac{P_j}{l_j}}$$
(3)

where  $(P_i/l)$  is the gas permeance of a membrane in GPU (1 GPU = 1 × 10<sup>-6</sup> cm<sup>3</sup>(STP)/cm<sup>2</sup> s cmHg), *I* represents the penetrating gas *i*,  $Q_i$  is the volumetric flow rate of gas permeated through the membrane (cm<sup>3</sup>/s, STP), *A* the effective membrane area (cm<sup>2</sup>),  $\Delta P$  is the transmembrane pressure (cmHg), and *T* is the temperature at which the permeation experiment being performed. Chromatograph (Perkin Elmer) was used to investigate the the composition of gases side.  $a_{ij}$  represents ideal selectivity of gas *i* to *j*.

## Experimental design

To extract mathematical model, find the most impact factor, the response surface methodology was used. If all variables are assumed measurable, the response surface can be expressed as follows:

$$y = f(x_1, x_2, ..., x_k)$$
(3)

Second order polynomial is usually considered as a full model in RSM[12-13]:

$$y = \beta_0 + \sum_{i=1}^k \beta_i x_i + \sum_{i=1}^k \beta_{ii} x_{ii}^2 + \sum_{i=1}^k \beta_{ij} x_i x_j + \varepsilon$$
(4)

where *y* is the response (CO<sub>2</sub> permeation),  $\beta$ s are regression coefficients, *x<sub>i</sub>* is a coded independent variable,  $\epsilon$  is the error and *k* is the number of factors. After optimizing membrane performance, the effect of stretch ratio(A), Zeolite content(B) and air gap distance (C) on membrane CO<sub>2</sub> and CH<sub>4</sub> permeability and selectivity was investigated at optimized condition.

# Simulation and optimization of process configurations

The simulation method adapted in current work was conducted in Aspen Hysis. The hollow fiber membrane



Fig. 2: Schematic diagram of the experimental set-up.

unit operation extension is comprised of two independent constituents, namely the ActiveX Server dynamic link library (DLL) and extension definition file (EDF). The adopted algorithm is robust, efficient, and converges at the stage cuts as high as 99.0%. The convergence was determined by an error function with a default composition calculation tolerance of 0.001%. The overall workflow was presented in Fig. 3.

The input parameters used for the simulation condition are summarized in Table 1. The gas processing cost (GPC) is calculated according to the procedure and fundamental assumptions outlined in Table 2 and the data obtained from process simulation. The proposed design configurations include single stage (SS), single stage with permeate recycle (SSPR), double stage with permeate recycle (DDPR), double stage with retentate recycle (DSRR), triple stage with retentate recycle (TSRR) and triple stage with permeate and retentate recycle (TSPRR).

# **RESULTSA AND DISCUSSION**

## Membrane characterization

To find the most suitable spinning conditions for SAPO-34/PES mixed matrix dope, the effects of air gap distance was first studied since this parameter can be easily manipulated throughout the process. SAPO-34/PES hollow fiber air gap distance varied from 10 to 25 cm. Fig. 5 shows the FESEM images of cross-sectional view of MMMs with different air gap distance. Referring to Fig. 5, macro-voids are produced in resultant SAPO-34/PES at the air gap distance of 5 cm and 10 cm. Table 3 records the gas permeation data for SAPO-34/PES hollow



Fig. 3: Overall solution procedure for the hollow fiber membrane module simulation[14].

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Membrane module specification	0.2032  m diameter (1 m × 18.5 m × 0.008 m channels)		
Feed flow	35 MMSCFD (991,095 std.m <sup>3</sup> /day)		
Feed inlet pressure and temperature (typical offshore natural gas)	30 bar and 25 °C		
Feed Composition(mole %)         10-60 CO2 , balance CH4, 0.05 C2H6, 0.05 C3H			
Permeate pressure (bar)	1.4		
CH <sub>4</sub> permeance	$4.5E-06 \text{ cm}^3 \text{ (STP) cm}^{-2} \text{ s}^{-1} \text{ cmHg}^{-1}$		
CO <sub>2</sub> /CH <sub>4</sub> selectivity	20		
Residue CO <sub>2</sub> composition <2% to meet pipeline specification			
Membrane Fiber characteristic(inner/outer diameter)	250/100		
Membrane Porosity	0.5		
Effective membrane thickness	1000 Å (3.937 × 10 <sup>-3</sup> mil)		
Membrane module diameter	15 in		

Table 1: Input parameters used for the simulation condition.



Fig. 4: Basic simulated model.

fibers spun. Referring to Table,  $CO_2$  and  $CH_4$  permeances were decreased while gas pair selectivities were increased on the increment of air-gap distance from 5 cm to 15 cm. The reduction in gases permeance could be because of reduction of non-selective voids from the selective layer of MMMs[15].

#### **RSM** results

The analysis of variance (ANOVA) for removal efficiency (Y) is represented in Table 4. The correlation coefficient ( $R^2$ ) is 0.99, which it is greater than 0.80, the cut-off for a model with good fit. The final regression model is presented in terms of its coded factors:

 $Y = 6.190 + 0.3314 \times A + 0.1577 \times B + 0.2728 \times C +$ 0.1669 × AB + 0.1846 × AC + 0.0511 × BC -0.1045 × A<sup>2</sup> - 0.6540 × B<sup>2</sup> - 0.4977 × C<sup>2</sup> Where: stretch ratio(A), Zeolite content(B) and air gap distance (C).

Figs.6 shows the interaction effect of factors on response parameter. Based on the Figure, SR has a positive effect on CO2 permeability. Increasing SR results in decreasing membrane thickness which leads to improving mass transfer rate. Permeance and selectivity of stretched fibers were exceptionally better relative to non-stretched fiber. Based on the Fig.6b SAPO-34 presence has a synergetic effect on CO2 permeability. SAPO-34 zeolite is one of the inorganic materials that has been extensively employed in the synthesis of MMM for CO<sub>2</sub> gas separation. The molecular structure of SAPO-34 with pore size around 0.38 nm makes it as a promising filler for  $CO_2/CH_4$  gas separation application since it is larger than  $CO_2$  gas kinetic diameter (0.33 nm) and similar with CH<sub>4</sub> kinetic diameter (0.38 nm). Fig.6c shows the effect of air

Total plant investment (TPI):	TPI=TFI+SC		
Membrane module cost (MC)	\$10/ft <sup>2</sup> (includes cost of membrane element)		
Installed compressor cost (CC)	$8650 \times (HP/\eta)^{0.82}$ (for processes with recycle streams)		
Fixed cost (FC)	MC+CC		
Base plant cost (BPC)	1.12 * FC		
Project contingency (PC)	0.20 * BPC		
Total facilities investment (TFI)	BPC+PC		
Start up cost (SC)	0.10 * VOM		
Annual variable operating and maintenance Cost(VOM):	VOM=CMC+LTI+DL+LOC+MRC+UC		
Contract and material maintenance cost(CMC)	0.05 * TFI		
Local taxes and insurance (LTI)	0.015 * TFI		
Direct Labor cost (DL) (based on 8 h/day per 25 MMSCFD of feed)	\$ 15/h		
Labor overhead cost (LOC)	1.15 * DL		
Membrane replacement costs (MRC)	\$ 3/ft <sup>2</sup> of membrane		
Utility cost (UC)	\$ 0.07/kw h		
Wellhead price of crude natural gas	\$ 2/MMBTU		
Heating value of natural gas	1066.8 MMBTU/MMSCF		
On stream factor (OSF)	96%		
Compressor efficiency $(\eta_{cp})$	0.8		

 Table 2: Gas processing cost (GPC).



Fig. 5: FESEM images of cross-sectional view of SAPO-34/PES with different air gap distance of; (a) 5 cm, (b) 10 cm, (c) 15.0 cm, (d)20 cm.

Air gap Distance(cm)	Permeance, GPU <sup>a</sup>		Selectivity	
	$CH_4$	CO <sub>2</sub>	CO <sub>2</sub> /CH <sub>4</sub>	
5	$5.60\pm0.16$	$95.30 \pm 13.08$	17.02	
10	$4.11\pm0.12$	$86.42 \pm 10.06$	21.02	
15	$2.87\pm0.08$	$73.15 \pm 1.08$	25.48	
20	$3.32\pm0.23$	$75.90\pm3.85$	22.86	

Table 3: Effect of various air gap distance on hollow fiber permeation properties.

<sup>a</sup>  $1 GPU = 1 \times 10^{-6} cm^3 (STP)/cm^2 s cmHg.$ 

Source	Sum of Squares	DF	Mean Square	F-value	p-value	
Model	12.26	9	1.36	44.88	< 0.0001	significant
А	1.50	1	1.50	49.41	< 0.0001	
В	0.3397	1	0.3397	11.19	0.0074	
С	1.02	1	1.02	33.48	0.0002	
AB	0.2228	1	0.2228	7.34	0.0220	
AC	0.2727	1	0.2727	8.98	0.0134	
BC	0.0209	1	0.0209	0.6889	0.4259	
A <sup>2</sup>	0.1575	1	0.1575	5.19	0.0459	
B <sup>2</sup>	6.16	1	6.16	203.04	< 0.0001	
C <sup>2</sup>	3.57	1	3.57	117.60	< 0.0001	
Residual	0.3036	10	0.0304			
Lack of Fit	0.2490	5	0.0498	4.56	0.0607	not significant
Pure Error	0.0546	5	0.0109			
Cor Total	12.56	19				· · · · · · · · · · · · · · · · · · ·

Table 4: ANOVA results.

gap distance on CO<sub>2</sub> permeability. In low air gap distance, radial flow of solvent is not fully established and it cannot hinder the diffusion of external coagulant. Therefore finger-like macro-voids on cross-section of SAPO-34/PES are eliminated. This results leads to decrease CO2 permeability inside the membrane.

Maximum permeability and the corresponding optimal conditions of variables were determined and the model was confirmed by some further experimental runs. Numerical optimization was done to find a maximum point for the desirability function by setting the values of SR, SAPO-34 content and air gap distance within their ranges and maximizing the permeability. The results listed in Table 5. The desirability value was found to be 0.93. This optimum condition was checked experimentally.

The results showed the permeability of  $6.67 \times 10^{-6}$  (cm<sup>3</sup>(STP)/cm<sup>2</sup>.cmHg.s). The high degree of agreement between the predicted optimum conditions and the repeated experimental results indicated that the CCD RSM could be employed as an effective and reliable tool for evaluation and optimization of the effects of parameters on the permeability.

#### CO<sub>2</sub>/CH<sub>4</sub> separation experiments

Fig. 7 present  $CO_2$  and  $CH_4$  permeabilities of the PES/SAPO-34 MMMs as a function of upstream pressure. As observed, the both permeabilities decreases with increment of pressure.

Effect of operating temperature from 303 to 343 K on permeability and selectivity of the PES/SAPO-34 MMMs





Fig. 6: 3D plot of factors effect on CO<sub>2</sub> permeability.

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are presented in Fig. 8. As operating temperature is increases, permeabilities increase, while selectivity decreases. For  $CO_2$  molecules, two different impacts are : solubility decreases with increasing temperature, while diffusivity increases with increasing temperature. Since the diffusivity is stronger function of temperature than solubility, permeability increases.

## Comparison with other membranes

Table 6 has gathered the recently published  $CO_2/CH_4$  separation performance of polysulfune-based MMM hollow fiber. Our developed SAPO34 MMM exhibited lower selectivity than that of other MMMs in the literature, but with a relatively high permeance.

## Simulation Results

Fig. 9 shows the effect of feed composition on  $CH_4$  recovery for all proposed configurations. It can be observed that the  $CH_4$  recovery is reducing with the increase of  $CO_2$  in the feed gas. The simulated results proved that the employing multiple stage systems leads to high methane recovery.

Fig. 10 shows the effect of feed composition on the total membrane area. It was found that the area rises with the  $CO_2$  composition increment in the feed until it reaches its maximum point. Then, more increasing results in decreasing in the membrane area requirement.

To find the best design, GPC has to be minimum with respect to the operating conditions. Fig. 11 shows the effect of feed composition on the GPC. Single stage systems without recycle (SS) shows a moderate GPC since no compressors are needed and the area required is small. DSRR and triple TSPRR show the maximum GPC due to the high compressor power and very large membrane area required. TSRR results relatively less GPC because of the improved methane recovery.

## CONCLUSIONS

In this work, the defects in poletherysulfone (PES)/silicoaluminophosphate (SAPO)-34 zeolite mixed matrix membrane was prepared by dry–wet spinning technique for the separation of CO<sub>2</sub>/CH<sub>4</sub> mixtures. In this regard, the synthesized PES/SAPO-34 mixed matrix membranes (MMMs) were characterized via FESEM analyses. It was found that air gap height gave the most influence on gas permeation properties. The effects of

SR	2.5		
SAPO-34 content	23.7%		
Air gap distance	18		
CO2 permeability	6.54×10 <sup>-6</sup> (cm <sup>3</sup> (STP)/cm <sup>2</sup> .cmHg.s)		

Table 5: Optimised conditions



Fig. 7: Effect of feed pressure on a:CO2 b:CH4 permeance of the MMMs.



Fig. 8: Effect of feed temperature on a:CO2 b:CH4 permeance c: selectivity of the PES/SAPO-34.

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Membrane	Permeance, GPU		Selectivity	Pof
	CH4	CO2	CO2/CH4	Kel.
PSF	2.52	78.11	31.05	[26]
PSF + 0.1% fumed silica	2.75	90.04	32.74	[26]
PSF + 2% carbon black	2.16	76.25	35.40	[27]
$PSF + 5\% \mu CX$	2.45	95.40	39.10	[28]
PSF	2.19	60.09	27.44	[16]
PSF + 0.05% Cloisite® 15A	1.41	56.25	40.26	[16]
ZIF-8/Pebax 1657/PES	52	542	16.1	[17]
NaY zeolite- Matrimid®5218 matrix			43.3	[18]
ABS/PVAc		40.41	4.93	[19]
Pebax 1657/ ZIF-8			20.4	[20]
NaX/PES		187.76	57.41	[21]
PES/SAPO-34	3.41	96.25	28.22	Present Study

Table 6: A comparison of PES/SAPO 34 hollow fiber mixed matrix membranes for CO<sub>2</sub>/CH<sub>4</sub> separation.

 $\mu CX = microporous \ carbon \ xerogel.$ 



16000 14000 Total membrane area (m<sup>2</sup>) 12000 10000 8000 6000 4000 2000 0+ 0 0.1 0.2 0.3 0.4 0.5 0.6 CO<sub>2</sub> feed composition

Fig. 9: Effect of feed composition on methane recovery.

Fig. 10: Effect of feed composition on total membrane area.

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Fig. 11: Effect of feed composition on gas processing cost.

modified feed temperature and feed pressure on permeation and  $CO_2/CH_4$  selectivity of membranes were investigated. The MMMs showed better performance than the neat PES membrane. The maximum selectivity achieved was 28.22 at 0.5 bar pressure and 35 °C. The minimum GPC is achieved by the double stage system with permeate recycle because of the high methane recovery and moderate power requirement for the configuration.

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